CAMBRIDGE INTERNATIONAL EXAMINATIONS

Cambridge International General Certificate of Secondary Education

MARK SCHEME for the October/November 2015 series

0653 COMBINED SCIENCE

0653/33 Paper 3 (Extended Theory), maximum raw mark 80

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Р	age	2		Syllabus	Paper
			Cambridge IGCSE – October/November 2015	0653	33
1	(a)	larç into cap			[3
	(b)		aller airway diameter ; sence of mucus obstructing flow ;		[2
	(c)	(i)	$12.5 - 5.8 = 6.7 \text{ (dm}^3/\text{min)}$; $\frac{6.7}{5.8} \times 100 = 115(\%) \text{ or } 116(\%)$; or $\frac{12.5}{5.8} = 215$; 215 - 100 = 115 (%);		[max 2
		(ii)	to get more oxygen (into the blood)/remove more carbon dioxide ; for $\underline{\text{respiration}}$;		[2
	(d)	mu airv bac or (mo airv mo	ia become paralysed) cus cannot be shifted upwards; vays become even more restricted/more mucus for bacteria to breed cteria/tar will not be removed from the lungs/increases risk of bronchi ore mucus is produced) vay becomes even more restricted/blocked; re mucus for bacteria to breed in/remain in lungs/increases chances est infections/reduces oxygen supply for the body;	tis ;	[max 2] [Total: 11]
2	(a)	the	rmal energy to chemical energy ;		[1]
	(b)	(i)	steeper gradient than solid line ;		[1]
		(ii)	increasing concentration increases rate of reaction; increased frequency of collisions;		[2]
	(c)	(i)	atoms ions ions ; (must be in this order)		[1]
		(ii)	silver written below copper;		[1]

[Total: 7]

[1]

(iii) the atoms of more reactive metals become ions more readily than those of

less reactive metals;

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3	(a)	wei	ght/gravitational force (accept gravity);		[1]
	(b)	(gra	avitational) potential (energy) to kinetic (energy) ;		[1]
	(c)	(i)	accelerating; constant speed;		[2]
		(ii)	area under graph between A and C or $(\frac{1}{2} \times 2 \times 9) + (2 \times 9)$ or $\frac{1}{2}(2 + 2) \times 9 \times 9 \times 9 = 27$ (m);	-4) × 9 ;	[2]
	(d)	d) particles far apart in air/gas, but close together/touching in water/liquid (accept diagrammatic description); stopping the skateboarder requires loss of KE; more work done/loses more KE/more difficult to push water particles aside than pushing air particles aside/owtte/diagrams; harder to push water molecules apart because of the forces between them/more particles;			[max 3] [Total: 9]
4	(0)	/:\	contains the correct proportions of putrients for an individual :		[4]
4	(a)	(i)	contains the correct proportions of nutrients for an individual;		[1]
		(ii)	to prevent scurvy/AVP;		[1]
	(b)	(i)	as temperature increases the amount of vitamin C decreases ;		[1]
		(ii)	rate/amount of decomposition/breakdown/disappearance increas temperature;	es with	[1]
		(iii)	temperature will vary in different parts of the world; this will affect the amount of vitamin C (in fruit before the experimen OR	nt) ;	
			amount of water given/contained in1 fruit may vary; this could affect concentration of the fruit juice; OR		
			different variety of orange/fruits vary genetically; (naturally) contains different amounts of vitamin C; AVPs;;		[max 2]
	(c)	boil	ing water destroy some/all of the <u>vitamin C</u> ;		[1]
	(d)	(advantage) convenient if you need to leave the baby / mother may not have enough milk/AVP;		•	
		•	eadvantage) es not contain antibodies / any reference to bonding / AVP ;		[2]
			, ,		r—1

Mark Scheme

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[Total: 9]

Syllabus

Paper

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5 (a)

A **←**

В ———

[1]

(b) (i)

observation	explanation
(bubbles of gas)	hydrogen;
(changes from green to purple)	alkaline solution ;

[2]

(ii) more vigorous reaction/hydrogen produced faster/greater temperature rise; alkali metals become more reactive down the group; (accept correct references to ease of ionisation of metal atoms)

[2]

[2]

(iii) one electron in shell; all Group I elements have 1 electron in outer shell;

[Total: 7]

6 (a) (i) $(\frac{10}{20} =) 0.5$;

Hz/hertz;

[2]

(ii)
$$(\frac{30}{10}) = 3 \text{ (m/s)}$$

[1]

(iii) $v = f\lambda$ (in any form)/ $\frac{3}{0.5}$;

= 6 (m);

(allow ecf from (i) and/or (ii))

[2]

(b) $2 \times \text{amplitude } (0.5) = 1 \text{ (m)};$

[1]

(c) R placed in right-hand end box;

[1]

(d) (i) tidal energy more predictable/nearer to land so easier access/installation/any other reasonable suggestion based on access to the energy involved or easier technology/ORA;

[1]

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(ii) (efficiency =
$$\frac{useful\ energy\ output}{energy\ input}$$
)
$$= \frac{150}{500} \times 100 = 30(\%);$$
 [1]

[Total: 9]

7 (a) allows solar radiation to reach the Earth's surface;

contains greenhouse gas molecules/example;

(greenhouse gases in the atmosphere) absorb infra-red radiation (that is emitted from the Earth's surface);

infra-red radiation is then (re-)emitted by the greenhouse gases into the atmosphere/ back to the Earth's surface;

[max 2]

(b) two from:

carbon dioxide/methane/water vapour;

[1]

(c) two from:

(increased) use of fossil fuels/example;

(increased) deforestation;

(increased) keeping of cows/growing rice;

AVP; [max 2]

(d) reduced use of fossil fuels/removal of carbon dioxide from exhaust/promote the use of public transport;

reduced deforestation;

reduced agricultural practices that cause methane to be produced;

plant more trees;

use more renewable energy sources;

AVP;

[Total: 6]

[max 1]

(a) (i) $2HCl + (CuCO_3) \rightarrow (CuCl_2) + CO_2 + H_2O$ 8 formulae; balanced;

[2]

(ii) limewater;

turns milky/cloudy etc.;

[2]

(b) (i) copper (deposit on cathode);

chlorine (gas at anode);

[2]

(ii) (copper ions):

move towards cathode/negative (electrode);

(chloride ions):

move towards anode/positive (electrode);

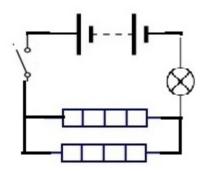
[2]

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- (c) (i) CuCl; reference to the need for charge balance; [2]
 - (ii) high density/high melting point/form coloured compounds/act as catalysts; [max 1]
- (d) correct structure of methane molecule ; correct structure of ethane molecule ; [2]

[Total: 13]

- 9 (a) (i) move towards each other; unlike charges attract; [2]
 - (b) (i) force; [1]
 - (ii) any path heading towards the upper positive plate; [1]
 - (c) (i) complete circuit with 2 extra components included in series and/or in parallel; two heaters in parallel; lamp in series in main circuit;



[3]

(ii) water expands/volume increases/particles get further apart water becomes less dense; (less dense)/warm water rises (above denser colder water)/owtte; [2]

[Total: 9]